

# NASA TECH BRIEF

## *Marshall Space Flight Center*



NASA Tech Briefs announce new technology derived from the U.S. space program. They are issued to encourage commercial application. Tech Briefs are available on a subscription basis from the National Technical Information Service, Springfield, Virginia 22151. Requests for individual copies or questions relating to the Tech Brief program may be directed to the Technology Utilization Office, NASA, Code KT, Washington, D.C. 20546.

### Study of In-Situ Degradation of Thermal Control Surfaces

A study has been made of the damage mechanism to semiconductor pigments exposed to ultraviolet radiation. The experimental technique used in this study is of broad interest for two reasons: it may be more widely used for investigations of surface chemistry, and it may be used analytically to determine contamination.

The materials investigated were binderless films of zinc oxide (ZnO), rutile titania ( $\text{TiO}_2$ ), and zinc orthotitanate ( $\text{Zn}_2\text{TiO}_4$ ). Each of these was exposed to ultraviolet radiation in a vacuum, causing changes in the reflective properties of the pigment. Since it was already known that damage to the pigments could be "bleached out" by air, test gases ( $\text{O}_2$ ,  $\text{CO}_2$ , and CO) were added in definite increments. This made it possible to study the reversal of the damage and draw conclusions about both the bleaching and the damaging of the pigments.

At each stage of the experiment, reflective spectra (from  $0.325 \mu\text{m}$  to  $2.6 \mu\text{m}$ ) were taken in situ, to determine how gas pressure affected adsorption. The ZnO and the  $\text{Zn}_2\text{TiO}_4$  exhibited a square root dependence on oxygen pressure, suggesting monatomic oxygen adsorption and desorption. Ultraviolet damage to  $\text{TiO}_2$

appeared to involve at least two types of defects; at least one of which was surface oxygen related. Carbon dioxide ( $\text{CO}_2$ ) did not seem to be involved in any of the surface reactions, but evidence suggested CO may react with adsorbed or lattice oxygen to form  $\text{CO}_2$ .

#### Note:

Additional information may be obtained from:  
Technology Utilization Officer  
Marshall Space Flight Center  
Code A&TS-TU  
Huntsville, Alabama 35812  
Reference: B72-10336

#### Patent status:

No patent action is contemplated by NASA.

Source: J. E. Gilligan and G. A. Zerlaut of  
IIT Research Institute  
under contract to  
Marshall Space Flight Center  
(MFS-20892)